Second-Generation Trifluoromethyl-Substituted Chiral Dendrimers Containing Triply Branched Building Blocks: CF₃ as Sensitive NMR Probe for 'Remote' Diastereotopicity ¹)

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(S)-4,4,4-Trifluoro-3-hydroxybutanoic acid was used as a starting material for the synthesis of dendritic branches (16-21, 30-32) which were attached to a chiral triol (6 or *ent*-6), derived from (R)-3-hydroxybutanoic acid, to give CF₃-substituted dendrimers (33-36, 38-40) and dendritic compounds (37) of 1st and 2nd generation. The key steps in these syntheses are diastereoselective aldol additions of dioxanone enolates (building blocks and intermediates 6-15, 22, 23, and 25-28) and *Williamson* etherifications of benzylic-branch bromides with triols (intermediates and products 16-21, 28, and 30-40). The surfaces of the dendrimers are covered with MeO (33-35, 38-40) or allyloxy groups (36 and 37). The new dendrimers are characterized by NMR and mass spectroscopy. ¹⁹F-NMR Signals of the CF₃ groups reveal constitutional heterotopicities caused by substituents which are separated from the F-nuclei by up to 15 bonds (through a 1,1'-biphenyl-4,4'-diyl spacer!) (Fig. 6).

1. Introduction. – There exist four fundamentally different ways for the construction of chiral dendrimers [2][3]. The first one consists of attaching branches to a chiral core; in a second one chiral building blocks are used as spacers or branching units, with an achiral core; in a third mode, the surface of a dendrimer from achiral building blocks is functionalized with chiral units; a fourth possibility is the attachment of achiral branches of different generations to a nonplanar core. The synthesis of chiral dendrimers, their properties as well as first applications have been reviewed in detail [3][4].

Triggered by our interest in chiral structures and enantioselective catalysis, we developed synthetic methods to build chiral dendrimers from non-racemic derivatives of 'tris(hydroxymethyl)methane' [5-7]. During these investigations, we discovered an unprecedented case of diastereoselectivity [8], and we have been able to synthesize dendrimers up to the fourth generation with doubly branching chiral building blocks [9]. Besides the primarily synthetic work accomplished so far, we have obtained first promising results from applications of our chiral dendritic systems [10][11].

With the main goal to gather additional structural information, we have now extended our work to an investigation of F-containing chiral dendrimers³). Our group has been engaged for many years in studies on chiral, CF₃-substituted compounds, because of their surprising reactivity, and with the aim to better understand the inter- and in-

¹⁾ Partially published in a preliminary communication [1].

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³⁾ For reviews on biologically active fluoro-organics, for applications in medicinal chemistry, as agrochemicals, and for fluorinated polymers which are important as thermoplastics, elastomers, membranes, and coatings, see [12-15].

tramolecular interactions of fluoro substituents in organic compounds [16][17]. The use of ¹⁹F as NMR probe was highly attractive as a tool for obtaining structural information about our chiral dendrimers.

To date, only few publications dealing with the synthesis of F-containing dendritic compounds have appeared. *Percec et al.* synthesized F-containing monodendra which were functionalized with crown ethers, and they observed the formation of nanotubes through self-organization due to the fluorophobic effect [18][19]. Surface functionalization of poly(propylenimine) dendrimers with perfluorinated alkanes by *Meijer et al.* led to so-called unimolecular inverse micelles [20]. From REDOR experiments on fluorinated ¹³C-labelled poly(benzyl ether) dendrimers, *Wooley et al.* concluded that the peripheral groups undergo an inward-folding, depending on the dendrimer size [21][22]. Molecular-modelling studies could then be performed on the basis of the measured ¹³C, ¹⁹F distances in these dendrimers.

Here, we report on the first synthesis of CF₃-containing *chiral* dendrimers and the investigations of their properties.

2. Synthesis of the Chiral Building Blocks and Branches. – The starting materials for the center pieces and branches of the new dendrimers are the carbonyl compounds 1-5 and the triol 6 [5][8]. The source of the CF₃-substituted components was the commercially available 4,4,4-trifluoro-3-oxobutanoate (1) which was treated with NaBH₄ and NaOH using known literature procedures [23][24]. Resolution of the resulting acid with 1-phenylethylamine afforded the enantiomerically pure (S)-4,4,4-trifluoro-3-hydroxybutanoic acid (2) [25], which was acetalized with pivalaldehyde to afford, after recrystallization from pentane, the pure cis-dioxanone 3 [26]. The Li-enolate of 3 was generated by deprotonation with t-BuLi as a base [16][27]. We first added this enolate to the benzaldehyde derivative 4: to our surprise, and in contrast to the nonselective (1.2:1) [8] reaction with the non-fluorinated analogue [5][28], we isolated two diastereoisomers 7 and 8 in

a 7:1 ratio (*Scheme 1*). The configuration of the major product 7 (2R,5R,6R,1'S) was determined by X-ray crystal-structure analysis (*Fig. 1*). This proves that the product has been formed by a (Re,Si) or *unlike* combination of the trigonal centers in the aldol addition⁴).

The primary Li-aldolate adduct may be depicted as shown in Fig. 2, a and b. We assume that the preference for the unlike combination of the trigonal centers results from a π -stacking interaction between the benzaldehyde and the enolate π -systems; this interaction appears to be stronger in the presence of a CF₃ group as compared to a CH₃ group in the 6-position of the Li-enolate. The relative topicity of the addition of the Li-enolate from 3 to aliphatic aldehydes is like (Fig. 2, c), i.e., opposite to that of the reaction with an aromatic aldehyde. It is interesting to note that, in the case of aliphatic aldehydes, the diastereoselectivities are identical for the CF₃- and CH₃-substitued dioxanones [5][16].

To prevent an observed *retro*-aldol reaction during the purification of the aldol product by flash chromatography, we used AcCl and pyridine instead of NH_4Cl for quenching of the aldol reaction at -78° . The resulting acetylated derivative 7 could be easily separated from the crude product mixture by precipitation with Et_2O . The reduction of 7 was performed with $NaBH_4$ in THF/MeOH 20:1 and furnished the triol 9 as a viscous oil in analytically pure form (after flash chromatography and rigorous removal of traces of solvent). Etherification with MeI and $CH_2 = CHCH_2Br$ afforded the peripheral building blocks 10 and 13, respectively, and subsequent treatment with Bu_4NF (TBAF) gave the benzyl alcohols 11 and 14, respectively, in almost quantitative yields.

⁴⁾ A control experiment with unsubstituted benzaldehyde afforded a major product (6:1) with the same relative configuration (established by X-ray crystal-structure analysis) as that of 7.

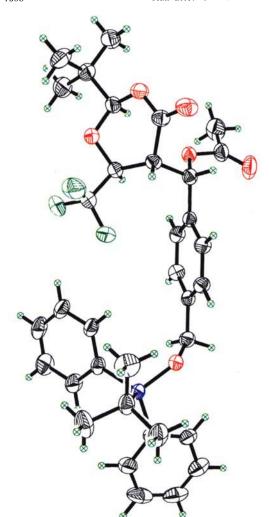


Fig. 1. ORTEP Representation of dioxanone 7 ((2R,5R,6S,1'S)-configuration). The thermal elipsoids are drawn to the 30% probability level. The X-ray structure was determined by Rheiner.

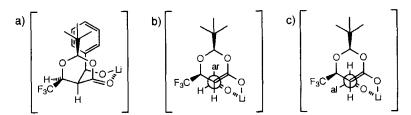


Fig. 2. Mechanism for the addol addition of the enolate of 3 to different aldehydes. a) Zimmermann-Traxler model and b) corresponding Newman projection for the addition to aromatic (ar) aldehydes, and c) to aliphatic (al) aldehydes.

Treatment with Ph₃P/Br₄C produced the corresponding 1st-generation benzyl bromides 12 and 15 in ca. 95% yield (Scheme 1).

The benzyl bromides 12 and 15 were then used for the etherification of the OH groups of the triol 9 under *Williamson* conditions (NaH, THF, reflux). After flash chromatography (Et₂O/pentane/CH₂Cl₂), the 2nd-generation products 16 and 19 were isolated in analytically pure form in *ca.* 60% yield (*Scheme 2*). Removal of the (*t*-Bu)Ph₂Si (TBDPS) protecting group (\rightarrow 17, 20) and treatment with Ph₃P/Br₄C yielded the 2nd-generation benzyl bromides 18 and 21, respectively.

The strategy described above has also been applied for the construction of branches containing 1,1'-biphenyl-4,4'-diyl spacers 5). Aldol addition of 3 to the TBDPS-protected biphenyl-carbaldehyde 5 afforded the diastereoisomeric dioxanones 22 and 23 (6:1). These isomers had to be separated by flash chromatography, during which partial β -elimination occurred (\rightarrow 24), probably due to the acidity of the used silica-gel material 6) 7). The resulting mixture 22/24 was treated with NaBH₄ in THF/MeOH 20:1 and furnished the triol 25 and the diol 29, which, fortunately, could be readily separated by flash chromatography. Etherification of the OH groups of the biphenyl derivative 25 (\rightarrow 26), deprotection with TBAF (\rightarrow 27), and bromination under the above mentioned conditions afforded the 1st-generation bromide 28. Coupling of the triol 25 with 28 gave

⁵⁾ The aldehyde 5 was prepared by Suzuki coupling of the 4-formylphenylboronic acid [29] with TBDPS-protected 4-bromobenzyl alcohol [8].

⁶) Treatment of the silica gel with Et₃N prior to chromatography resulted in *complete* β-elimination.

⁷⁾ With the non-fluorinated hydroxybenzylated dioxanones, we have never encountered such problems [5][28].

the TBDPS-protected branch 30, which, in turn, gave the 2nd-generation branch bromide 32 (60% overall yield) after treatment with TBAF (\rightarrow 31) and Ph₃P/Br₄C.

All of the above described building blocks and dendritic branches were fully characterized (¹H-, ¹³C-, and ¹⁹F-NMR, IR, MS (EI, FAB, MALDI-TOF) spectra as well as elemental analysis) and were shown to be very pure.

3. Preparation and Characterization of the CF_3 -Substituted Chiral Dendrimers 33–36 and 38–40. – The center pieces in our dendrimer synthesis are the enantiomeric triols 6 and *ent*-6, the coupling of which with the benzyl bromides was achieved by treating the triol with excess NaH (9 equiv.) in refluxing THF and adding an excess of the corresponding branched benzyl bromide (3.3–4 equiv.).

The coupling of 6 and *ent-*6 with the benzyl bromide 12 gave, after flash chromatography, the 1st-generation chiral dendrimers 33 and 34 (*Fig. 3*), respectively, in *ca.* 70% yield. While no significant differences are observed in the ¹H-NMR spectra of the diastereoisomeric compounds 33 and 34, ¹⁹F-NMR spectroscopy seems to be an ideal

means for detecting CF₃-group diastereotopicity. In the spectrum of 33, the three CF₃ groups are isochronous and give rise to a *doublet*; in compound 34, however, with the core from 6 exchanged by its enantiomeric counterpart from *ent*-6, the CF₃ groups 'perceive' the difference, and three of the possible six signals are seen.

The analysis of the 19 F-NMR spectrum of the 2nd-generation dendrimer **35** (*Fig. 4*), containing twelve CF₃ groups or 36 F-atoms, and having a molecular weight of 3702 Da, exhibits a clear-cut difference between the inner CF₃ groups and those located at the periphery (*Fig. 4*). Moreover, the three inner CF₃ groups of **35** give rise to four of the six possible signals, while eight signals can be recognized from the nine constitutionally heterotopic outer CF₃ groups 8).

During the synthesis of the dendrimers containing allyl groups at the periphery, we did not encounter any problems with the 1st-generation chiral dendrimer 36 (Fig. 5), except that the yield of the Williamson-coupling reaction was lower (40% compared to the 72% obtained with bromide 12). However, the triol 6 reacted only twice with the 2nd-generation branched benzyl bromide 21 to afford the dendritic compound 37. Thus, the replacement of peripheral CH_3 by allyl groups prevents the 'acceptance' of a third branch'). This case of incomplete coupling probably results from steric hindrance of the 'third' OH groups (cf. the diastereoisomeric differentiation described in our previous paper [8]). This is especially remarkable, because we are not dealing with 'real' dendrimers of spherical shape and with steric congestion at the periphery. The periphery of compounds 36 and 37 is ready for chemical modifications, for instance, by hydroboration or hydrosilylation of the allyl groups.

Similar observations of differences between analogous nuclei in various *layers* of dendrimers have been made in the ¹H-NMR spectra of the non-fluorinated dendrimers, where the signals of the peripheral, interior, and central Me groups give rise to distinct sets of signals [8].

As in the non-fluorinated 'doubly coupled' derivatives of this type (see Fig. 2 and Fig. 4 in [8]), there is a distinct upfield shift of the central CH resonance in the ¹H-NMR spectrum of 37, as compared to the 'fully coupled' dendrimers.

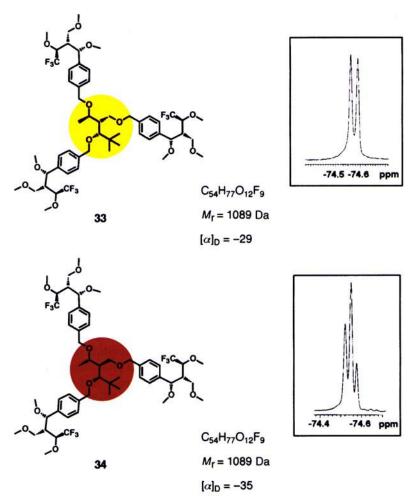
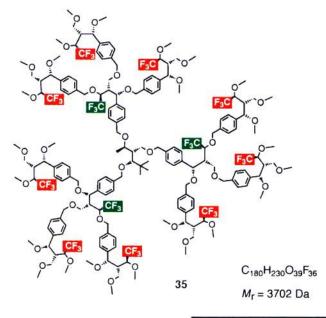
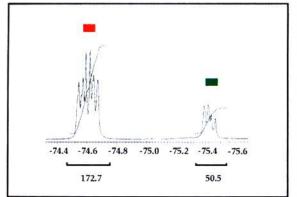


Fig. 3. Formulae and ¹⁹F-NMR spectra of 1st-generation dendrimers 33 and 34 (CDCl₃). Note that the two dendrimers differ only by having enantiomeric center piece units.

To reduce steric hindrance, we synthesized the dendrimers 38-40 containing biphenyl moieties. The 1,1'-biphenyl-4,4'-diyl system appeared to be the most suitable 'spacer'. By coupling the branch bromide 28 with the center pieces 6 and *ent-6*, we obtained the 1st-generation dendrimers 38 and 39, respectively. We now also succeeded in preparing the 2nd-generation dendrimer 40 (M_r 4614 Da) (Fig. 6). Inspite of the greatly increased distances between the CF_3 groups and the heterotopicity-generating constitutional differences in these 'elongated' dendrimers, three signals from the inner and four signals from the outer CF_3 group can be recognized in the ^{19}F -NMR spectrum (282-MHz spectrometer) 10).

¹⁰) ¹⁹F-NMR Measurements with 40 at different temperatures (-35° to 25°) did not reveal the existence of preferred axially chiral biphenyl conformations in this type of structure.





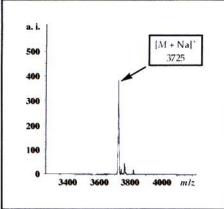


Fig. 4. Formula, ¹⁹F-NMR (CDCl₃), and MALDI-TOF mass (α-cyano-4-hydroxycinnamic-acid (CCA) matrix) spectra of 2nd-generation dendrimer 35

The dendrimers 33–36 and 38–40 as well as the dendritic compound 37 are very viscous oils or glassy solids, well soluble in solvents of varying polarity (toluene, Et₂O, CH₂Cl₂). They have been characterized by ¹H-, ¹³C-, and ¹⁹F-NMR, and IR spectroscopy, and elemental analysis. Since the accuracy of these methods is not sufficient (due to the oligomeric nature of the dendritic macromolecules) to detect subtle structural differences, *i.e.*, structural defects, the purity of the compounds was demonstrated by MALDI-TOF mass spectrometry (see *Figs. 4–6*).

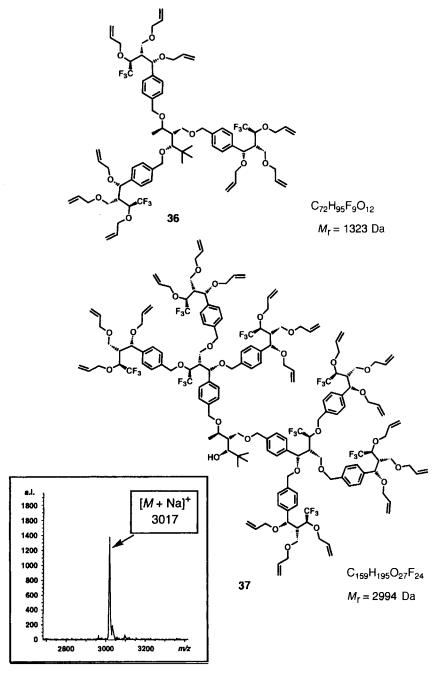


Fig. 5. Formulae of the 1st- and 2nd-generation dendrimers 36 and 37 with peripheral allyl groups, and MALDI-TOF mass spectrum of compound 37 (α-cyano-4-hydroxycinnamic-acid (CCA) matrix)

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Experimental Part

- 1. General. All reactions were carried out under Ar. Reagent-grade chemicals were purchased from Fluka or Aldrich, and used without further purification unless otherwise stated. Crude solvents for chromatography and for workup were distilled from Sikkon (Et₂O from KOH/FeSO₄), THF was freshly distilled from sodium benzophenone ketyl radical. The compounds 1-6 and ent-6 were prepared according to literature procedures [8][16][25][29]. TLC: glass-plated TLC silica gel 60 F_{2.54} (Merck). Flash chromatography (FC): silica gel 60 (Merck) 40-63 mm. M.p. Büchi 510, uncorrected. Optical rotations: Perkin-Elmer-241 polarimeter, 10-cm cells; in CHCl₃. IR: Perkin-Elmer-1600-FTIR, in cm^{-1. 1}H-, ¹³C-, and ¹⁹F-NMR: Bruker AMX-II-500, AMX-400, AMX-300, Varian-Gemini-200, and -300 spectrometers at r.t.; in CDCl₃: chemical shifts, δ, in ppm downfield from internal TMS, coupling constants, J, in Hz. MS (m/z (%)): Hitachi-Perkin-Elmer RMU-6M for EI; VG-ZAB2-SEQ for FAB in a 3-nitrobenzyl-alcohol matrix (3-NOBA). Matrix-assisted laser desorption ionization time-of-flight (MALDI-TOF) on a Bruker-ReflexTM instrument with a N₂ laser system (337 nm), positive-ion mode. α-cyano-4-hydroxycinnamic acid (CCA); spectra were processed and printed using the X-MASS program on a SUN workstation. Elemental analyses were performed by the Mikroanalytisches Laboratorium der ETH-Zürich. Abbreviations: TBDPS-Cl ((t-Bu)Ph₂SiCl), LAH (LiAlH₄), TBAF (Bu₄NF · 3 H₂O), r.e. (rotary evaporator).
- 2. Nomenclature. The nomenclature used for identification of the dendrimers, dendritic compounds, and dendritic branches is based on the convention proposed by Seebach and coworkers (see Fig. 8 in [8]).
- (2R.5R.6S)-2-(tert-Butyl)-5- $[(S)-(4-\{[(\text{tert-butyl})diphenylsityloxy]methyl\}phenyl)acetoxymethyl]-6-(tri-fluoromethyl)-1,3-dioxan-4-one (7). A soln. of 6.0 g (26.5 mmol) of 3 in THF (45 ml) was cooled to <math>-78^{\circ}$, and 19.5 ml (29.2 mmol) of t-BuLi were added at such a rate that the inner temp. never exceeded -70° , and the mixture was subsequently kept at -78° for 20 min. Compound 4 (11.9 g, 31.8 mmol) in THF (45 ml) was slowly added, the mixture kept at -78° for 6 h, then quenched with 3.8 ml (53 mmol) of AcCl and 10.7 ml (132.5 mmol) of pyridine. After stirring at r.t. for 16 h, the mixture was extracted with 10% HCl, H₂O, and sat. NaCl (70 ml each), the combined org. extracts were dried (MgSO₄) and concentrated *in vacuo*. ¹⁹F-NMR of the crude product showed a ratio 7/8 of 7:1. The main isomer 7 was precipitated with Et₂O (11.1 g, 65%). Upon slow recrystallization from

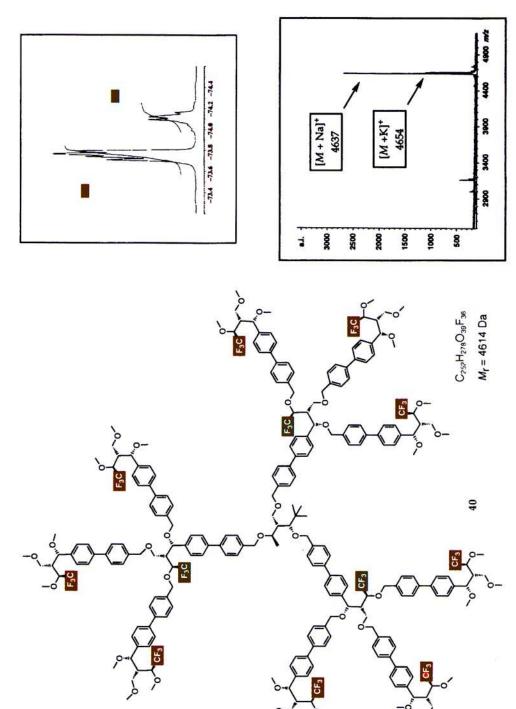


Fig. 6. Formula, 19 F-NMR (CDCl3), and MALDI-TOF mass (anthracene-1,8,9-triol matrix) spectra of 2nd-generation dendrimer 40

Et₂O, crystals of 7, suitable for X-ray analysis were isolated. M.p. $168-169^{\circ}$. [α]_D^{T,L} = -24.9 (c=1.6). IR: 2963m, 2932m, 2859m, 1758s, 1428m, 1371m, 1278m, 1150s, 1110s, 1010m. ¹H-NMR (400 MHz): 0.96 (s, t-Bu-C(2)); 1.09 (s, t-BuSi); 2.18 (s, AcO); 3.43 (dd, J=5.6, 4.7, H-C(5)); 4.43-4.49 (m, H-C(6)); 4.74 (s, H-C(2)); 4.76 (s, C H_2 OSi); 6.22 (d, J=4.7, H-C(1')); 7.26-7.68 (m, 14 arom. H). ¹³C-NMR (100 MHz): 19.31; 20.93; 23.54; 26.82; 34.94; 45.72; 65.00; 72.04; 74.68; 105.26; 122.71; 126.09; 126.54; 127.77; 129.79; 133.31; 133.85; 135.53; 142.44; 165.45; 168.73. ¹⁹F-NMR (282 MHz): -79.56 (d, J=6.0). EI-MS: 585 (18, $[M-57]^+$), 439 (43), 395 (17), 360 (53), 317 (29), 221 (33), 201 (62), 199 (100), 197 (75), 177 (51), 119 (34), 91 (84), 57 (60), 43 (56), 41 (37). Anal. calc. for $C_{35}H_{41}F_3O_6Si$ (642.50): C 65.40, H 6.43; found: C 65.60, H 6.51.

 $(18,2R,3S)-1-(4\cdot\{[(\text{tert}-Butyl)diphenylsityloxy]methyl\}phenyl)-2-(hydroxymethyl)-4,4,4-trifluorobutane-1,3-diol (9). To NaBH₄ (0.59 g, 15.6 mmol) in THF (30 ml)/MeOH (3 ml) at 0°, 7 (2 g, 3.1 mmol) in THF (30 ml) was added. The mixture was stirred at r.t. for 7 h, then diluted with H₂O (30 ml) and saturated with NaCl. The mixture was extracted with Et₂O (3 × 70 ml), the combined org. extracts were dried (MgSO₄) and concentrated in vacuo. The crude product was purified by FC (pentane/Et₂O 1:1): 1.4 g (87%) of 9. Colorless, very viscous oil. [a]<math>_{5}^{\text{th}} = -16.0$ (c = 1.9). IR: 3386m, 2931m, 2859m, 1601w, 1472m, 1428s, 1274s, 1169s, 1113s, 1074s, 1017m. $_{1}^{\text{th}}$ H-NMR (400 MHz): 1.10 (s, t-Bu); 2.08 – 2.11 (m, H – C(2)); 2.14 (br. s, OH); 2.65 (d, J = 3.1, OH); 3.60 – 3.63 (m, 1 H, CH_{2} OH); 4.11 – 4.14 (m, 1 H, CH_{2} OH); 4.24 (d, J = 7.1, OH); 4.57 – 4.61 (m, H – C(3)); 4.78 (s, CH_{2} OSi); 5.22 – 5.24 (m, H – C(1)); 7.35 – 7.70 (m, 14 arom. H). $_{1}^{13}$ C-NMR (100 MHz): 19.34; 26.87; 45.23; 61.48; 65.23; 70.22; 72.84; 125.15; 126.21; 126.44; 127.75; 129.77; 133.42; 135.58; 140.25; 141.31. $_{1}^{19}$ F-NMR (282 MHz): $_{1}^{19}$ H-Of.89 (d, J = 8.0). EI-MS: 461 (9, $[M - 57]^{+}$), 365 (10), 317 (10), 245 (7), 231 (12), 211 (7), 199 (70), 197 (11), 195 (13), 183 (16), 129 (15), 91 (100), 77 (18), 57 (16). Anal. calc. for $C_{28}H_{33}F_{3}O_{4}$ Si (518.64): C 64.84, H 6.41; found: C 64.88, H 6.51.

 $\begin{array}{l} 1-\{[(\text{tert-}Butyl)\, diphenylsityloxy\,] methyl\}-4-\{[(S,2R,3S)-4,4,4-trifluoro-1,3-dimethoxy-2-(methoxymethyl)-butyl\,] butyl\,] benzene (\textbf{10}). \ To \ NaH (0.48 g, 20 mmol) in THF (10 ml) at 0°, a soln. of \textbf{9} (1.15 g, 2.2 mmol) in THF (20 ml) was added, and the mixture was stirred at r.t. for 30 min, prior to the addition of MeI (1.25 ml, 20 mmol). The mixture was heated to reflux for 3 h, then quenched with <math>H_2O$ (40 ml). The mixture was extracted with Et_2O (3 × 40 ml), and the combined org. extracts were dried (MgSO₄) and concentrated *in vacuo*. The crude product was purified by FC (pentane/Et₂O 9:1): 1.13 g (91 %) of pure 10. Colorless, viscous oil. [a]_{D}^{t.t.} = -17.5 (c = 1.8). \\ \text{IR: } 2932m, 2859m, 1463w, 1428w, 1382w, 1279m, 1170s, 1112s, 1018w. }^{11} \text{H-NMR} (300 \text{ MHz}): 1.10 (s, t-Bu); \\ 2.14-2.20 (m, H-C(2)); 3.12 (s, MeO); 3.23 (s, MeO); 3.23-3.35 (m, CH₂OMe); 3.82 (s, MeO); 4.12 (qd, J = 8.0, H-C(3)); 4.37 (d, J = 8.3, H-C(1)); 4.79 (s, CH₂OSi); 7.25-7.72 (m, 14 arom. H). $^{13}\text{C-NMR}$ (75 MHz): 19.33; 26.87; 46.02; 56.78; 58.64; 61.31; 65.36; 68.27; 77.04; 80.88; 126.07; 127.25; 127.70; 127.83; 129.71; 133.53; 135.60; 138.22; 140.85. $^{19}\text{F-NMR}$ (282 MHz): -74.52 (d, J = 7.8). EI-MS: 503 (4, [M - 57] +), 389 (7), 352 (27), 351 (100), 241 (59), 231 (22), 213 (15), 209 (48), 201 (19), 199 (13), 166 (43), 134 (32), 129 (35), 91 (42), 57 (2). Anal. calc. for $C_{31}H_{39}F_{3O}A_{5}$ i (564.70): C 66.40, H 7.01; found: C 66.46, H 6.70.

4-[(1S,2R,3S)-4,4,4-Trifluoro-1,3-dimethoxy-2-(methoxymethyl))butyl]benzenemethanol (11). The compound 10 (1.08 g, 1.94 mmol) in THF (25 ml) was treated with TBAF (1.22 g, 3.88 mmol) for 12 h at r.t., then $\rm H_2O$ (15 ml) was added. The combined org. extracts were dried (MgSO₄) and concentrated in vacuo. The residue was purified by FC (pentane/Et₂O 1:1): 0.55 g (89%) of pure 11. Colorless, viscous oil. [α]_D¹: = -35.3 (c = 1.6). IR: 3460w, 3008m, 2932s, 2823w, 1608w, 1499w, 1453w, 1383m, 1272s, 1170s, 1120s, 1021s. ¹H-NMR (300 MHz): 1.81 (br. s, OH); 2.13 – 2.19 (m, H–C(2)); 3.11 (s, MeO); 3.22 (s, MeO); 3.21 – 3.45 (m, CH₂OMe); 3.61 (s, MeO); 4.10 (qd, J = 8.0, 2.3, H–C(3)); 4.38 (d, J = 8.2, H–C(1)); 4.72 (d, J = 5.0, CH₂OH); 7.31 – 7.40 (m, 4 arom. H). ¹³C-NMR (75 MHz): 45.97; 56.81; 58.66; 61.33; 65.10; 68.22; 77.22; 80.80; 125.02; 127.09; 127.61; 139.16; 140.59. ¹⁹F-NMR (282 MHz): -74.57 (d, J = 7.8). EI-MS: 322 (1, M^+), 307 (1), 259 (13), 229 (2), 215 (1), 177 (5), 152 (23), 151 (100), 147 (4), 44 (7). Anal. calc. for C₁₅H₂₁F₃O₄ (322.32): C 55.90, H 6.57; found: C 55.86, H 6.79.

I-(Bromomethyl)-4-[(18,2R,38)-4,4,4-trifluoro-1,3-dimethoxy-2-(methoxymethyl)hutyl]henzene (12). To a soln. of 11 (1.45 g, 4.52 mmol) in THF (60 ml) at 0°, Ph₃P (2.37 g, 9.03 mmol) and Br₄C (3.0 g, 9.03 mmol) were added. After stirring at r.t. for 16 h, H₂O (60 ml) was added. The mixture was extracted with CH₂Cl₂ (3 × 100 ml), dried (MgSO₄), and evaporated. The crude product was purified by FC (pentane/Et₂O 6:1): giving 1.39 g (80 %) of pure 12. Colorless, viscous oil. [α]_D^{t,t} = -26.5 (c = 2.2). IR: 3005m, 2933m, 2825m, 1512m, 1448m, 1384m, 1279m, 1170s, 1145s, 1103s. ¹H-NMR (300 MHz): 2.10 – 2.17 (m, H – C(2)); 3.11 (s, MeO); 3.21 – 3.46 (m, CH₂OMe); 3.23 (s, MeO); 3.61 (s, MeO); 4.08 (gd, J = 8.0, 2.3, H – C(3)); 4.38 (d, J = 8.2, H – C(1)); 4.51 (s, CH₂Br); 7.29 – 7.42 (m, 4 arom. H). ¹³C-NMR (75 MHz): 33.18; 45.99; 56.93; 58.62; 61.34; 68.17; 77.14; 80.75; 127.80; 127.90; 129.16; 137.45; 140.20. ¹⁹F-NMR (282 MHz): -74.56 (d, J = 8.2). EI-MS: 384 (< 1, M⁺), 305 (7), 273 (33), 215 (91), 214 (11), 213 (100), 135 (15), 134 (41), 119 (13), 91 (19), 45 (15). Anal. calc. for C₁₅H₂₀BrF₃O₃ (385.22): C 46.77, H 5.23; found: C 46.95, H 5.23.

 $\begin{array}{l} 1-(Bromomethyl)-4-\{(18,2R,38)-1.3-diallyloxy-2-(allyloxymethyl)-4,4,4,-trifluorobutyl\}benzene~(\textbf{15}).~\text{As described for }\textbf{12},~\text{with }\textbf{14}~(0.43~\text{g},~1.08~\text{mmol}),~\text{Ph}_3P~(0.42~\text{g},~17.36~\text{mmol}),~\text{and }\text{Br}_4C~(0.53~\text{g},~1.62~\text{mmol}),~\text{reaction time }16~\text{h.}~\text{FC}~(\text{pentane/Et}_2O~18:1):~\textbf{15}~(0.45~\text{g},~90~\%).~\text{Slightly-yellow, viscous oil.}~[\alpha]_{D}^{\text{r.f.}}=-~22.0~(c=1.6).~\text{IR}:~3082w,~3006w,~2868m,~1647w,~1422m,~1351m,~1279m,~1171s,~1092s,~996m,~933s.~^1\text{H-NMR}~(400~\text{MHz}):~2.24-2.29~(m,\text{H}-\text{C(2)});~3.39~('dd',~ABX,J=10.2,~6.2,~1~\text{H.}~CH_2\text{OAllyl});~3.51~('dd',~ABX,J=10.2,~3.9,~1~\text{H.}~CH_2\text{OAllyl});~3.68-4.31~(m,~3~\text{CH}_2=\text{CHC}H_2,~\text{H}-\text{C(3)});~4.50~(s,~\text{CH}_2\text{Br});~4.56~(d,J=7.4,~\text{H}-\text{C(1)});~5.05-5.35~(m,~3~\text{CH}_2=\text{CHCH}_2);~5.66-5.98~(m,~3~\text{CH}_2=\text{CHCH}_2);~7.29-7.52~(m,~4~\text{arom.}\text{H.}).~^{13}\text{C-NMR}~(100~\text{MHz}):~33.17;~46.04;~65.83;~70.02;~71.75;~73.91;~74.63;~78.83;~116.50;~116.95;~117.57;~126.64;~127.75;~129.12;~134.10;~134.39;~134.66;~137.35;~140.41.~^{19}\text{F-NMR}~(282~\text{MHz}):~-75.00~(d,J=10.9).~\text{EI-MS}:~463~(4,M^+),~422~(19),~407~(100),~327~(19),~255~(10),~241~(35),~239~(36),~229~(12).~\text{Anal. calc. for $C_{21}\text{H}_{26}\text{BrF}_3\text{O}_3~(463.33):~C~54.44,~\text{H}~5.66;~\text{found}:~C~54.47,~\text{H}~5.75.} \end{array}$

 $(MeO)_9$ - $\{(S)G_2(F_{12})\}^3$ -OTBDPS (16). To NaH (0.40 g, 16.74 mmol) in THF (20 ml) at 0° , a soln of 9 (0.96 g, 1.86 mmol) in THF (10 ml) was added, and the mixture was stirred at r.t. for 30 min. The bromide 12 (2.86 g, 7.44 mmol) in THF (20 ml) was added at 0°, and the mixture was stirred under reflux for 19 h. The reaction was quenched with H_2O (50 ml), and the mixture was extracted with Et_2O (3 × 100 ml). The combined org. extracts were dried (MgSO₄) and concentrated in vacuo. The crude product was purified by FC (CH₂Cl₂/pentane/Et₂O 6:3:0.5): 1.63 g (62%) of pure 16. Slightly-yellow, viscous oil. [α]_D^{r.t.} = -40.6 (c = 1.4). IR: 3005w, 2931m, 1463w, 1428w, 1384w, 1279m, 1170s, 1141s, 1104s, 1019w. ¹H-NMR (400 MHz): 1.11 (s, t-Bu); 2.13-2.15 $(m, 3 \text{ H}-\text{C}(2)(\text{G}_2)); 2.42-2.44 \quad (m, \text{H}-\text{C}(2)(\text{G}_1)); 3.08-3.11 \quad (3s, 3 \text{ MeO(P)}); 3.19-3.21 \quad (3s, 3 \text{ MeO(P)});$ 3.22-3.36 (m, $3 CH_2OMe(P)$); 3.59-3.60 (2s, 3 MeO(P)); 3.57-3.69 (m, $OCH_2-C(2)(G_1)$); 4.05-4.13 $(m, 3 \text{ H}-\text{C}(3)(\text{G}_2)); 4.21-4.76 \ (m, \text{H}-\text{C}(3)(\text{G}_1), \ 3 \text{ OC}\\ H_2\text{Ph}, \ 3 \text{ H}-\text{C}(1)(\text{G}_2), \ \text{H}-\text{C}(1)(\text{G}_1)); 4.80 \ (s, \text{CH}_2\text{OSi});$ 7.18-7.71 (m, 26 arom. H). ¹³C-NMR (100 MHz): 19.35; 26.88s, 46.04; 46.33; 56.75; 56.83; 58.62; 61.30; 65.33; 66.49; 68.21; 70.90; 72.72; 74.74; 77.06; 79.55; 80.81; 124.39; 126.20; 127.21; 127.29; 127.35; 127.39; 127.51; 127.70; 127.74; 127.86; 129.75; 133.47; 135.60; 137.48; 137.90; 138.15; 138.26; 138.86; 139.29; 139.40; 141.00. ¹⁹F-NMR (282 MHz): -74.52 to -74.63 (m, 3 CF₃(G₂)); -75.19 (d, J = 7.9, CF₃(G₁)). FAB-MS: 1431 (1, M^+), 1259 (1), 305 (8), 274 (16), 273 (100), 241 (12), 229 (51), 197 (24). Anal. calc. for C₇₃H₉₀F₁₂O₁₃Si (1431.55): C 61.25, H 6.34; found: C 61.09, H 6.46.

 $(MeO)_{\circ}-[(S)G_2(F_{12})]^3$ -OH (17). As described for 11, with 16 (1.63 g, 1.15 mmol) and TBAF (0.72 g, 2.28 mmol), reaction time 16 h. FC (pentane/Et₂O 1:1): 17 (1.1 g, 81%). Slightly-yellow, viscous oil. [a]_D^{t,t} = -50.1 (c=1.8). IR: 3450w, 3005w, 2933m, 2882m, 2820m, 1513w, 1461w, 1420w, 1385m, 1277s, 1169s, 1138s, 1102s. ¹H-NMR (400 MHz): 1.76 (t, t = 5.9, OH); 2.14-2.16 (t = t

 $\begin{array}{l} {\rm H-C(1)(G_1));\ 4.72\ (\textit{d},\textit{J}=5.8,\textit{C}H_2OH);\ 7.15-7.52\ (\textit{m},\ 16\ {\rm arom.\ H}).\ ^{13}C-{\rm NMR}\ (100\ {\rm MHz});\ 46.04;\ 46.28;\ 56.77;\ 56.85;\ 58.63;\ 61.31;\ 65.02;\ 66.39;\ 68.21;\ 70.86;\ 72.62;\ 74.75;\ 77.20;\ 79.63;\ 80.80;\ 127.19;\ 127.25;\ 127.37;\ 127.41;\ 127.61;\ 127.72;\ 127.86;\ 137.39;\ 137.75;\ 138.08;\ 138.82;\ 139.15;\ 139.36;\ 139.47;\ 140.79.\ ^{19}F-{\rm NMR}\ (282\ {\rm MHz});\ -74.51\ {\rm to}\ -74.63\ (\textit{m},\ 3\ {\rm CF}_3(G_2));\ -75.02\ (\textit{d},\textit{J}=8.1,\ {\rm CF}_3(G_1)).\ {\rm FAB-MS}:\ 1192\ (4,\textit{M}^+),\ 1160\ (8),\ 1021\ (16),\ 643\ (17),\ 503\ (8),\ 305\ (35),\ 273\ (100),\ 229\ (55),\ 134\ (38).\ {\rm Anal.\ calc.\ for\ C_{57}H_{72}F_{12}O_{13}\ (1193.17);\ C\ 57.38,\ H\ 6.08;\ found:\ C\ 57.31,\ H\ 6.07. \end{array}$

 $(MeO)_0 - [(S)G_2(F_{12})]^3 - Br$ (18). As described for 12, with 17 (0.83 g, 0.83 mmol), Ph₃P (0.32 g, 1.24 mmol), and Br₄C (0.41 g, 1.24 mmol). After stirring at r.t. for 16 h, another 1.5 equiv. Ph₃P/Br₄C was added and the mixture was stirred at r.t. for 24 h. After workup, FC (pentane/Et₂O 2:1) gave 18 (0.90 g, 86%) as a colorless oil. [2]_D^{1.1} = -45.2 (c = 1.4). IR: 3005w, 2933m, 2882w, 2820w, 1461w, 1385w, 1277s, 1169s, 1138s, 1103s. ¹H-NMR (400 MHz): 2.14-2.16 (m, 3 H-C(2)(G₂)); 2.43-2.45 (m, H-C(2)(G₁)); 3.08-3.11 (3s, 3 MeO(P)); 3.19-3.21 (3s, 3 MeO(P)); 3.24-3.33 (m, 3 CH₂OMe(P)); 3.55 ($^{\prime}$ dd', $^{\prime}$ ABX, $^{\prime}$ J = 10.2, 6.4, 1 H, OCH₂-C(2)(G₁)); 3.59-3.61 (3s, 3 MeO(P)); 3.65 ($^{\prime}$ dd', $^{\prime}$ ABX, $^{\prime}$ J = 10.2, 4.0, 1 H, OCH₂-C(2)(G₁)); 4.03-4.19 (m, 3 H-C(3)(G₂)); 4.22-4.76 (m, H-C(3)(G₁), 3 OCH₂Ph, 3 H-C(1)(G₂), H-C(1)(G₁)); 4.51 ($^{\prime}$ s, CH₂Br); 7.14-7.52 (m, 16 arom. H). ¹³C-NMR (100 MHz): 33.05; 46.03; 56.85; 58.63; 61.31; 68.22; 71.00; 72.70; 74.75; 77.20; 79.51; 80.79; 127.29; 127.37; 127.42; 127.47; 127.71; 127.81; 127.86; 129.29; 137.34; 137.61; 137.99; 138.90; 139.42; 140.20. ¹⁹F-NMR (282 MHz): -74.53 to -74.64 (m, 3 CF₃(G₂)); -75.11 (d, J = 7.8, CF₃(G₁)). FAB-MS: 1256 (8, M +), 1224 (7), 1086 (15), 1085 (30), 1084 (12), 1083 (28), 305 (36), 273 (100), 230 (15), 229 (72), 198 (22), 160 (32). Anal. calc. for C₅₇H₇₁BrF₁₂O₁₂ (1256.07): C 54.51, H 5.70; found: C 54.49, H 5.75.

(Allyl) $_9$ -[(S) $G_2(F_{12})$] 3 -OTBDPS (19). As described for 16, with 9 (1.12 g, 2.16 mmol), NaH (0.47 g, 19.44 mmol), and 15 (3.5 g, 7.56 mmol), reaction time 19 h. FC (pentane/Et₂O 10:1): 19 (1.48 g, 40%). Slightly-yellow, viscous oil. [α]_D^{1.1} = -38.5 (c=1.4). IR: 3008m, 2930m, 2860m, 1646w, 1512w, 1458w, 1427m, 1351w, 1262s, 1171s, 1139s, 1094s, 1019m. ¹H-NMR (400 MHz): 1.11 (s, t-Bu); 2.23–2.28 (m, 3 H–C(2)(G₂)); 2.43–2.44 (m, H–C(2)(G₁)); 3.36–4.78 (m, 3 CH₂OAllyl, OCH₂–C(2)(G₁), 3 H–C(3)(G₂), 9 CH₂=CHCH₂, H–C(3)(G₁), 3 OCH₂Ph, 3 H–C(1)(G₂), H–C(1)(G₁)); 4.80 (s, CH₂OSi); 5.02–5.36 (m, 9 CH₂=CHCH₂); 5.67–5.98 (m, 9 CH₂=CHCH₂); 7.16–7.71 (m, 26 arom. H). ¹³C-NMR (100 MHz): 19.35; 26.89; 46.12; 46.35; 65.34; 65.91; 66.51; 69.91; 69.99; 70.86; 771.74; 71.76; 71.77; 72.09; 72.70; 73.90; 74.23; 74.52; 74.83; 77.22; 78.87; 79.69; 116.38s, 116.40; 116.44; 116.77; 116.83; 117.47; 117.51; 124.22; 127.23; 127.27; 127.33; 127.43; 127.49; 127.71; 127.74; 127.78; 127.90; 127.75; 133.48; 134.17; 134.19; 134.48; 134.55; 134.74; 134.78; 135.60; 137.40; 137.82; 137.86; 138.05; 138.27; 139.08; 139.46; 139.51; 139.66; 141.02. ¹⁹F-NMR (282 MHz): -74.25 to -74.40 (m, 4 CF₃). MALDI-TOF-MS: 1688 ([M + 23] $^+$). Anal. calc. for C₉₁H₁₀₈F₁₂O₁₃Si (1665.91): C 65.61, H 6.53; found: C 65.38, H 6.30.

(Allyl) $_{9}$ -[(S) $G_{2}(F_{12})$] 3 -OH (20). As described for 11, with 19 (1.38 g, 0.83 mmol) and TBAF (0.52 g, 1.66 mmol), reaction time 16 h. FC (pentane/Et $_{2}$ O 6:4): 20 (0.95 g, 81%). Slightly-yellow, viscous oil. [x] $_{0}^{\text{Ft}}$: = -51.3 (c = 1.4). IR: 3684w, 3619m, 3008s, 2976s, 2888s, 1646w, 1514w, 1477w, 1422m, 1351w, 1262s, 1171s, 1140s, 1094s, 1046s. 1 II-NMR (400 MHz): 1.74 (t, J = 5.9, OH): 2.23–2.29 (m, 3 H–C(2)(G $_{2}$)); 2.43–2.44 (m, H–C(2)(G $_{1}$)); 3.37–4.77 (m, 3 CH $_{2}$ OAllyl, OCH $_{2}$ –C(2)(G $_{1}$), 3 H–C(3)(G $_{2}$), 9 CH $_{2}$ =CHCH $_{2}$, H–C(3)(G $_{1}$), 3 OCH $_{2}$ Ph, 3 H–C(1)(G $_{2}$), H–C(1)(G $_{1}$)); 4.71 (d, J = 5.8, CH $_{2}$ OH); 5.03–5.36 (m, 9 CH $_{2}$ =CHCH $_{2}$); 5.67–5.99 (m, 9 CH $_{2}$ =CHCH $_{2}$); 7.10–7.37 (m, 16 arom. H). 13 C-NMR (100 MHz): 46.09; 46.13; 46.31; 65.01; 65.92; 66.40; 69.93; 70.00; 70.84; 71.76; 71.78; 72.60; 73.91; 74.46; 74.65; 74.85; 75.32; 75.60; 78.85; 78.87; 79.80; 116.42; 116.45; 116.82; 116.85; 117.52; 124.22; 127.04; 127.18; 127.34; 127.36; 127.67; 127.74; 127.80; 134.17; 134.20; 134.47; 134.56; 134.75; 134.77; 137.32; 137.68; 137.97; 139.05; 139.15; 139.58; 139.74; 140.82. 19 F-NMR (282 MHz): -74.15 to -74.40 (m, 4 CF $_{3}$). MALD1-TOF-MS: 1450 ([m + 23] $^{+}$). Anal. calc. for C_{75} H $_{90}$ F $_{12}$ O $_{13}$ (1427.51): C 63.10, H 6.35; found: C 63.25, H 6.32.

 $(Allyl)_{9} - [(S)G_{2}(F_{12})]^{3} - Br \ (\textbf{21}). \text{ As described for } \textbf{12}, \text{ with } \textbf{20} \ (0.86 \text{ g}, 0.61 \text{ mmol}), \text{ Ph}_{3}\text{P} \ (0.24 \text{ g}, 0.91 \text{ mmol}), \text{ and } \text{Br}_{4}\text{C} \ (0.30 \text{ g}, 0.91 \text{ mmol}). \text{ After stirring at r.t. for } 16 \text{ h}, \text{ another } 1.5 \text{ equiv. of } \text{Ph}_{3}\text{P}/\text{Br}_{4}\text{C} \ \text{ was added and the mixture was stirred at r.t. for } 24 \text{ h}. \text{ After workup, } \text{FC (pentane/Et}_{2}\text{O} \ 8.5:1.5) \text{ gave } \textbf{21} \ (0.82 \text{ g}, 90\%). \text{ Colorless oil. } [\textbf{z}]_{9}^{\text{Int}} = -37.8 \ (c = 1.6). \text{ IR: } 3008s, 2872m, 1646w, 1513w, 1458w, 1422m, 1351w, 1278s, 1171s, 1140s, 1094s, 1019m. $^{1}\text{H}-\text{NMR}$ \ (400 \text{ MHz}): $2.23-2.30$ \ (m, 3 \text{ H}-\text{C(2)}(\text{G}_{2})); $2.41-2.42$ \ (m, \text{H}-\text{C(2)}(\text{G}_{1})); $3.38-4.76$ \ (m, 3 \text{ CH}_{2}\text{OAllyl}, \text{OCH}_{2}-\text{C(2)}(\text{G}_{1}), 3 \text{ H}-\text{C(3)}(\text{G}_{2}), 9 \text{ CH}_{2}-\text{CHCH}_{2}, \text{H}-\text{C(3)}(\text{G}_{1}), 3 \text{ OCH}_{2}\text{Ph}, 3 \text{ H}-\text{C(1)}(\text{G}_{2}), \text{H}-\text{C(1)}(\text{G}_{1})); $4.51 \ (s, \text{CH}_{2}\text{Br}); 5.03-5.36$ \ (m, 9 \text{ CH}_{2}-\text{CHCH}_{2}); 5.67-5.98$ \ (m, 9 \text{ CH}_{2}-\text{CHCH}_{2}); 7.10-7.42$ \ (m, 16 \text{ arom. H}). $^{13}\text{C-NMR}$ \ (100 \text{ MHz}): $3.302; 46.10; 46.27; 65.90; 66.32; 69.94; 69.99; 70.93; 71.74; 72.65; 73.88; 74.46; 74.63; 74.76; 75.32; 75.60; 78.82; 79.67; 116.38; 116.42; 116.78; 116.82; 117.46; 117.50; 124.22; 127.02; 127.18; 127.32; 127.33; 127.41; 127.71; 127.76; 127.86; 129.25; 129.85; 134.14; 134.19; 134.44; 134.54; 134.72; 134.77; 137.24; 137.52; 137.61; 137.84; 139.10; 139.62; 139.74; 140.17. $^{19}\text{F-NMR}$ \ (282 \text{ MHz}): -74.23 to -74.42 \ (m, 4 \text{ CF}_{3}). \text{ MALDI-TOF-MS}: 1513 \ ([M+23]^{1}). \text{ Anal. calc. for } \text{C}_{75}\text{H}_{89}\text{BF}_{12}\text{O}_{13}$ \ (1490.41): C \ 60.44, H \ 6.02; found: C \ 60.21, H \ 5.97.}$

(2R,5R,6S)-2-(tert-Butyl)-5-(4'-{f(tert-Butyl)diphenylsilyloxy]methyl}-1,f'-biphenyl-4-yl)(acetoxy)methyl)-6-(trifluoromethyl)-1,3-dioxan-4-one (22). As described for 7, with 0.84 g (3.7 mmol) of 3, 2.7 ml (4.7 mmol) of t-BuLi, 2 g (4.44 mmol) of 5, 0.53 ml (7.4 mmol) of AcCl, and 1.5 ml (18.5 mmol) of pyridine (22/23 ratio 5:1). FC (pentane/Et₂O 1:3) afforded 1.18 g (44%) of 22 with ca. 6% traces of 24 (according to 19 F-NMR). M.p. 62-65°. [x] $_{\rm L}^{\rm EL}$ = -29.3 (c = 1.6). IR: 2964m, 2923m, 2861m, 1759s, 1600w, 1425m, 1369m, 1276m, 1148s, 1112s, 1087m. 11 H-NMR (400 MHz): 0.96 (s, t-Bu-C(2)); 1.11 (s, t-BuSi); 2.21 (s, AcO); 3.47 (dd, J = 5.6, 4.9, H-C(5)); 4.47-4.51 (m, H-C(6)); 4.75 (s, H-C(2)); 4.81 (s, CH₂OSi); 6.25 (d, J = 4.8, H-C(1')); 7.36-7.72 (m, 18 arom. H). 13 C-NMR (100 MHz): 19.33; 20.93; 23.52; 26.85; 34.95; 45.70; 65.23; 72.11; 74.57; 105.30; 122.67; 126.51; 126.65; 126.93; 127.62; 127.74; 129.73; 133.45; 134.07; 135.57; 140.78; 142.12; 165.37; 168.79. 19 F-NMR (282 MHz): -78.78 (d, d) = 7.0). FAB-MS: 717 (20, d)-d), 661 (46), 575 (7), 529 (25), 463 (100), 273 (25), 197 (7), 135 (12). Anal. calc. for C $_{41}$ H $_{45}$ F $_{30}$ O $_{6}$ Si (718.90): C 68.50, H 6.31; found: C 68.65, H 6.38.

(1S,2R,3S)-1-(4'-{[(tert-Butyl) diphenylsityloxy]methyl}-1,1'-biphenyl-4-yt)-2-(hydroxymethyl)-4,4,4-trifluorobutane-1,3-diol (25). As described for 9, with 22 (1.29 g, 1.79 mmol) and NaBH₄ (0.34 g, 8.95 mmol). FC (pentane/Et₂O 1:1): 25. Slightly-yellow, viscous oil. [α]₅^{1.5} = -15.6 (c = 1.6). IR: 3415w, 2933m, 2851m, 15941w, 1497m, 1430s, 1272s, 1169s, 1138s, 1113s, 1077s. ¹H-NMR (400 MHz): 1.11 (s, t-Bu), 2.11-2.20 (m, H-C(2), OH); 2.70 (br. s, OH); 3.66-3.75 (m, 1 H, CH₂OH); 4.15-4.18 (m, 1 H, CH₂OH); 4.25 (d, J = 6.9, OH); 4.55-4.70 (m, H-C(3)); 4.82 (s, CH₂OSi); 5.30 (d, J = 7.7, H-C(1)); 7.36-7.73 (m, 18 arom. H). ¹³C-NMR (100 MHz): 19.36; 26.87; 45.19; 61.51; 65.29; 70.26; 72.72; 125.55; 126.51; 126.79; 126.91; 127.43; 127.76; 129.74; 133.48; 135.60; 139.06; 140.50; 141.12. ¹⁹F-NMR (282 MHz): -76.13 (d, J = 8.7). FAB-MS: 593 (18, [M - H])⁺, 577 (23), 537 (85), 339 (100), 291 (17), 196 (12), 135 (12). Anal. calc. for C₃₄H₃₇F₃O₄Si (594.74): C 68.66, H 6.27; found: C 68.37, H 6.24.

4'-[(1S,2R,3S)-4,4,4-Trifluoro-1,3-dimethoxy-2-(methoxymethyl)butyl]-1,1'-biphenyl-4-methanol (27). As described for 11, with 26 (1.42 g, 2.23 mmol) and TBAF (1.41 g, 4.46 mmol), reaction time 12 h. FC (pentane/Et₂O 1:1): 27 (0.83 g, 94%). Slightly-yellow, viscous oil. [α]_D^{r.t.} = -24.0 (c = 1.8). IR: 3459w, 3005m, 2931m, 2824w, 1602w, 1497w, 1462w, 1383m, 1278s, 1170s, 1120s, 1020s. ¹H-NMR (400 MHz): 1.80 (br. s, OH); 2.18–2.23 (m, H–C(2)); 3.13 (s, MeO); 3.26 (s, MeO); 3.27–3.44 (m, CH₂OMe); 3.63 (s, MeO); 4.14 (gd, J = 8.0, 2.3, H–C(3)); 4.41 (d, J = 8.3, H–C(1)); 4.74 (s, CH₂OH); 7.38–7.63 (m, 8 arom. H). ¹³C-NMR (100 MHz): 46.01; 56.88; 58.68; 61.37; 65.07; 68.27; 77.24; 80.81; 124.98; 127.07; 127.22; 127.49; 127.87; 130.08; 138.82; 140.04; 140.11; 140.39. ¹⁹F-NMR (282 MHz): -73.72 (d, J = 7.0). FAB-MS: 398 (14, M), 381 (44), 367 (25), 305 (43), 291 (20), 227 (100), 210 (14), 154 (15), 136 (16). Anal. calc. for C₂₁H₂₅F₃O₄ (398.42): C 63.31, H 6.32; found: C 63.35, H 6.32.

4-(Bromomethyl)-4'-[(1S,2R,3S)-4,4,4-trifluoro-1,3-dimethoxy-2-(methoxymethyl)butyl]-1,1'-biphenyl (28). As described for 12, with 27 (0.72 g. 1.79 mmol), Ph₃P (0.71 g. 2.69 mmol), and Br₄C (0.90 g. 2.69 mmol), reaction time 16 h. FC (pentane/Et₂O 9:1): 28 (0.75 g. 90%). White solid. M.p. 77–78°. [α]₀^{rd.} = -21.4 (c=1.7). IR: 3007m, 2933m, 2825w, 1609w, 1497m, 1462w, 1384w, 1279m, 1170s, 1141s, 1103s, 1066m, 1034w. ¹H-NMR (400 MHz): 2.17–2.22 (m, H—C(2)); 3.12 (s, MeO); 3.26 (s, MeO); 3.27–3.43 (m, C H_2 OMe); 3.62 (s, MeO); 4.13 (gd, J=8.0, 2.3, H—C(3)); 4.41 (d, J=8.3, H—C(1)); 4.55 (s, CH₂Br); 7.39–7.60 (m, 8 arom. H). ¹³C-NMR (100 MHz): 33.34; 46.03; 56.90; 58.68; 61.38; 68.26; 77.34; 80.80; 125.32; 127.45; 127.91; 129.55; 130.07; 136.90; 139.15; 139.98; 140.88. ¹⁹F-NMR (282 MHz): -73.73 (d, J=8.7). FAB-MS: 460 (14, $[M-H]^+$), 429 (32), 381 (82), 305 (96), 289 (100), 210 (51), 195 (44), 165 (21), 154 (55), 136 (43). Anal. calc. for C₂₁H₂₄BrF₃O₃ (461.32): C 54.68, H 5.24; found: C 54.73, H 5.37.

 $(MeO)_{9}$ - $[(S)G_{2}(F_{12})(Biph.)]^{3}$ -OTBDPS (30). As described for 16, with 25 (1.24 g, 2.08 mmol), NaH (0.45 g, 18.75 mmol), and 28 (3.36 g, 7.29 mmol), reaction time 19 h. FC $(CH_{2}Cl_{2}/pentane/Et_{2}O 6:3:0.5)$: 30 (2.93 g, 81%). Slightly-yellow, viscous oil. $[\alpha]_{0}^{\text{LL}} = -62.3 (c = 1.9)$. IR: 3008s, 2975s, 2894s, 1520s, 1475m, 1426m, 1391m, 1248s, 1170s, 1141s, 1103s, 1046s. ¹H-NMR (400 MHz): 1.14 (s, t-Bu); 2.18-2.22 (m, 3 H-C(2)(G₂));

 $(MeO)_9 - [(S)G_2(F_{12})(Biph.)]^3 - OH$ (31). As described for 11, with 30 (2.4 g, 1.39 mmol) and TBAF (0.87 g, 2.77 mmol), reaction time 16 h. FC (pentane/Et₂O 6:4): 31 (1.82 g, 89%). Colorless, viscous oil. [α]₀^{L.} = -74.1 (c = 1.5). IR: 3682w, 3619s, 3464w, 3009s, 2975s, 2932m, 2895m, 1520w, 1476w, 1423w, 1248s, 1140m, 1103s, 1046s. ¹H-NMR (400 MHz): 1.82 (t, t =

 $(MeO)_9 - [(S)G_2(F_{12})(Biph.)]^3 - Br$ (32). As described for 12, with 31 (1.43 g, 0.96 mmol), Ph₃P (0.38 g, 1.44 mmol), and Br₄C (0.48 g, 1.44 mmol). After stirring at r.t. for 16 h, another 1.5 equiv. of Ph₃P/Br₄C was added, and the mixture was stirred at r.t. for 24 h. After workup, FC (pentane/Et₂O 2:1) gave 32 (1.2 g, 80%). Slightly-yellow, viscous oil. $[a]_{D}^{15} = -60.4$ (c = 1.4). IR: 3008s, 2975s, 2894w, 2434m, 1520s, 1476m, 1423s, 1103s, 1046s. ¹H-NMR (400 MHz): 2.19-2.22 (m, 3 H-C(2)(G₂)); 2.48-2.52 (m, H-C(2)(G₁)); 3.11-3.12 (3s, 3 MeO(P)); 3.25-3.27 (3s, 3 MeO(P)); 3.28-3.37 (m, 3 CH₂OMe(P)); 3.58 (* $^{\prime}$ dd*, $^{\prime}$ ABX, $^{\prime}$ J = 10.0, 6.3, 1 H, OCH₂-C(2)(G₁)); 3.62-3.63 (2s, 3 MeO(P)); 3.73 (* $^{\prime}$ dd*, $^{\prime}$ ABX, $^{\prime}$ J = 10.1, 3.6, 1 H, OCH₂-C(2)(G₁)); 4.10-4.82 (m, 3 H-C(3)(G₂), H-C(3)(G₁), 3 OCH₂Ph, 3 H-C(1)(G₂), H-C(1)(G₁)); 4.55 (s, CH₂Br); 7.16-7.62 (m, 32 arom. H). 13 C-NMR (100 MHz): 33.27; 46.05; 46.37; 56.88; 58.67; 61.36; 66.34; 68.27; 70.59; 72.64; 74.56; 75.10; 79.14; 80.82; 125.30; 126.81; 126.98; 127.04; 127.05; 127.07; 127.24; 127.49; 127.83; 127.87; 127.92; 128.06; 128.36; 128.61; 129.23; 129.58; 136.72; 137.00; 137.11; 137.47; 138.72; 138.75; 139.21; 139.71; 140.12; 140.27; 140.31; 140.42; 140.82. 19 F-NMR (282 MHz): -73.72 to -73.79 (m, 3 CF₃(G₂)); -73.91 (d, J = 8.8, CF₃(G₁)). FAB-MS: 1573 (12, [M + 13]*), 1390, (10), 1251 (7), 781 (15), 381 (100), 305 (32), 210 (24), 195 (22), 139 (10). Anal. calc. for C₈₁H₈₇BrF₁₂O₁₂ (1560.46): C 62.35, H 5.62; found: C 62.23, H 5.42.

(*MeO*)₉-{[(S)G₁(F₉)]³}₃-{C} (33). To NaH (50 mg, 2.1 mmol) in THF (7 ml), **6** (40 mg, 0.23 mmol) in THF (6 ml) was added, and the mixture was stirred under reflux for 30 min. The bromide **12** (335 mg, 0.92 mmol) in THF (7 ml) was then added at 0°, and the mixture was stirred under reflux for 4 h. The reaction was quenched by the addition of H₂O (20 ml), and the mixture was extracted with Et₂O (3 × 40 ml). The combined org. extracts were dried (MgSO₄) and concentrated *in vacuo*. The crude product was purified by FC (pentane/Et₂O 4:1 → 1:1): **33** (180 mg, 74%). Colorless, glassy oil. [α]_D¹⁻¹ = − 29.1 (c = 1.6). IR: 2936m, 1704w, 1464w, 1383m, 1279m, 1170s, 1140s, 1103s, 1020m. ¹H-NMR (400 MHz): 0.94 (s, t-Bu); 1.31 (d, J = 6.4, Me); 2.05 – 2.15 (m, 3 H – C(2)(G₁)); 2.31 – 2.34 (m, H – C(3)(C)); 3.09 – 3.11 (3s, 3 MeO(P)); 3.20 – 3.22 (3s, 3 MeO(P)); 3.22 – 3.33 (m, 3 CH₂OMe(P), H – C(4)(C)); 3.59 – 3.62 (m, 1 H, CH₂(C), 3 MeO(P)); 3.74 – 3.83 (m, 1 H, CH₂(C), H – C(2)(C)); 4.05 – 4.14 (m, 3 H – C(3)(G₁)); 4.34 – 4.37 (m, 3 H – C(1)(G₁)); 4.38 – 4.68 (m, 3 OCH₂Ph); 7.24 – 7.38 (m, 12 arom. H). ¹³C-NMR (75 MHz): 16.67; 26.34; 37.42; 44.93; 46.03; 56.66; 56.79; 58.64; 61.32; 68.26; 70.70; 72.83; 73.64; 77.46; 80.85; 86.40; 125.70; 126.93; 127.15; 127.34; 127.51; 127.67; 127.83; 127.95; 138.39; 138.68; 138.80; 139.39. ¹⁹F-NMR (282 MHz): −74.58 (br. s, 3 CF₃). FAB-MS: 1087 (5, M *), 1056 (2), 917 (6), 457 (9), 425 (12), 305 (57), 273 (100), 229 (43), 198 (13), 134 (36), 91 (10). Anal. calc. for C₅₄H₇₇F₉O₁₂ (1089.17): C 59.55, H 7.13; found: C 59.57, H 7.19.

 $(MeO)_9 - \{[(S)G_1(F_9)]^3\}_3 - [C_{enant.}]$ (34). As described for 33, with *ent-6* (40 mg, 0.23 mmol), NaH (50 mg, 2.1 mmol), and 12 (335 mg, 0.92 mmol), reaction time 4 h. FC (pentane/Et₂O 4:1 \rightarrow 1:1): 34 (175 mg, 70%). [2l_D^{LL} = -35.0 (c = 1.8). IR: 3005w, 2933m, 2831w, 1461w, 1385w, 1277m, 1169s, 1138s, 1103s. ¹H-NMR (400 MHz): 0.94 (s, t-Bu); 1.31 (d, J = 6.4, Me); 2.09–2.16 (m, 3 H–C(2)(G₁)); 2.31–2.33 (m, H–C(3)(C));

3.09-3.11 (3s, 3 MeO(P)); 3.21-3.22 (3s, 3 MeO(P)); 3.22-3.32 (m, 3 CH₂OMe(P), H–C(4)(C)); 3.59-3.63 (m, 1 H, CH₂(C), 3 MeO(P)); 3.73-3.83 (m, 1 H, CH₂(C), H–C(2)(C)); 4.06-4.12 (m, 3 H–C(3)(G₁)); 4.33-4.37 (m, 3 H–C(1)(G₁)); 4.41-4.68 (m, 3 OCH₂Ph); 7.24-7.38 (m, 12 arom. H). ¹³C-NMR (100 MHz): 16.62; 26.33; 37.40; 44.90; 46.02; 56.78; 58.62; 61.31; 68.24; 68.35; 70.63; 72.81; 73.63; 76.49; 76.90; 80.84; 86.34; 126.95; 127.14; 127.33; 127.63; 127.77; 138.41; 138.68; 138.79; 138.86; 139.35. ¹⁹F-NMR (282 MHz): -74.52 to -74.58 (m, CF₃). FAB-MS: 1089 (10, M^+), 1087 (42), 918 (12), 917 (24), 306 (10), 305 (46), 273 (100), 229 (49), 160 (19), 135 (31), 134 (45). Anal. calc. for C₄₄H₇₇F₉O₁₂ (1089.18): C 59.55, H 7.13; found: C 59.56, H 7.09.

 $(MeO)_{27}$ -{ $[(S)G_2(F_{36})]^3\}_3$ -{[C]} (35). As described for 33, with 6 (33 mg, 0.19 mmol), NaH (41 mg, 1.71 mmol), and 18 (830 mg, 0.67 mmol), reaction time 24 h. FC (pentane/Et₂O 1:1): 35 (397 mg, 58%). [α]_D^{LL} = -47.3 (c = 1.6). IR: 3301m, 2934m, 1710w, 1513w, 1463w, 1383m, 1279m, 1170s, 1141s, 1103s, 1019m. ¹H-NMR (400 MHz): 0.96 (s, t-Bu); 1.36 (d, J = 6.4, Me); 2.12-2.17 (m, 9 H-C(2)(G₂)); 2.40-2.43 (m, 3 H-C(2)(G₁), H-C(3)(C)); 3.07-3.11 (4s, 9 MeO(P)); 3.17-3.21 (7s, 9 MeO(P)); 3.22-3.32 (m, 9 CH₂OMe(P), H-C(4)(C)); 3.59-3.69 (3s, 9 MeO(P), 3 OCH₂-C(2)(G₁), 1 H, CH₂(C)); 3.85-3.87 (m, H-C(2)(C)); 3.90-3.92 (m, 1 H, CH₂(C)); 4.04-4.13 (m, 9 H-C(3)(G₂)); 4.23-4.78 (m, 3 H-C(3)(G₁), 9 H-C(1)(G₂), 3 H-C(1)(G₁), 12 OCH₂Ph); 7.20-7.47 (m, 48 arom. H). ¹³C-NMR (100 MHz): 16.30; 26.43; 37.56; 44.52; 45.53; 46.06; 46.20; 46.37; 56.77; 56.82; 56.83; 58.59; 58.61; 61.29; 66.47; 68.21; 68.90; 70.64; 71.04; 72.85; 73.66; 74.70; 74.85; 75.08; 75.31; 75.55; 76.16; 79.38; 79.41; 80.78; 80.81; 80.84; 85.74; 124.68; 126.74; 126.91; 126.94; 127.06; 127.19; 127.21; 127.30; 127.33; 127.40; 127.50; 127.67; 127.87; 128.03; 137.43; 137.79; 137.84; 138.13; 138.59; 138.75; 138.86; 138.91; 138.92; 138.96; 139.37; 139.39. ¹⁹F-NMR (282 MHz): -74.52 to -74.58 (m, 3 CF₃(G₂)); -75.39 to -75.50 (m, CF₃(G₁)). MALDI-TOF: 3724 ([m + Na] +). Anal. calc. for C₁₈₀H₂₃₀F₃₆O₃₉ (3701.72): C 58.40, H 6.26; found: C 58.15, H 6.09.

($Allyl_{9}$ -{ $\{(S)G_{1}(F_{9})\}^{3}\}_{3}$ -{ $C\}$ } (36). As described for 33, with 6 (257 mg, 1.46 mmol), NaH (315 mg, 13.12 mmol), and 15 (2.7 g, 5.83 mmol), reaction time 4 h. FC (pentane/Et₂O 8:1): 36 (0.75 g, 40%). [α]_D^{T,L} = -20.7 (c=2.0). IR: 2869 κ , 1423 κ , 1352 κ , 1278m, 1170 κ , 1140 κ , 1095 κ , 1018m. ¹H-NMR (400 MHz): 0.94 (s, t-Bu); 1.30 (d, J=6.4, Me); 2.25–2.29 (m, 3 H–C(2)(G₁)); 2.30–2.34 (m, H–C(3)(C)); 3.22 (d, J=1.9, H–C(4)(C)); 3.34–4.69 (m, 3 CH₂OAllyl, CH₂(C), 9 CH₂=CHCH₂, 3 H–C(3)(G₁), H–C(2)(C), 3 OCH₂Ph, 3 H–C(1)(G₁)); 5.02–5.39 (m, 9 CH₂=CHCH₂); 5.67–5.99 (m, 9 CH₂=CHCH₂); 7.20–7.35 (m, 12 arom. H). ¹³C-NMR (100 MHz): 16.56; 26.30; 26.47; 37.40; 44.82; 46.09; 65.90; 68.39; 69.93; 70.66; 71.72; 71.75; 72.80; 73.58; 73.88; 74.55; 76.48; 78.82; 78.85; 78.93; 86.18; 116.39; 116.72; 116.77; 117.45; 124.19; 126.86; 127.21; 127.58; 127.73; 129.85; 134.18; 134.53; 134.55; 134.57; 134.77; 138.56; 138.59; 138.66; 139.01; 139.07; 139.23. ¹⁹F-NMR (282 MHz): -74.28 to -74.37 (m, 3 CF₃). MALDI-TOF: 1362 ([M + K]⁺), 1346 ([M + Na]⁺). Anal. calc. for C_{72} H₉ $_{85}$ F₉O₁₂ (1323.52): C 65.34, H 7.23; found: C 65.29, H 6.98.

(Allyl)₁₈-{[(S)G₂(F₂₄)]³}₂-{C}-OH (37). As described for 33, with 6 (27 mg, 0.16 mmol), NaH (33 mg, 1.36 mmol), and 21 (790 mg, 0.53 mmol), reaction time 24 h. FC (pentane/Et₂O 8:1): 37 (278 mg, 58%). [α]_D^{1,1} = -45.3 (c=1.4). IR: 3683m, 3620s, 3449s, 3008s, 2975s, 2892s, 2433s, 1521s, 1476s, 1422s, 1248s, 1140s, 1047s. ¹H-NMR (500 MHz): 0.92 (s, t-Bu); 1.35 (d, J=6.3, Me); 2.02–2.03 (m, H–C(3)(C)): 2.25–2.29 (m, 6 H–C(2)(G₂)); 2.41–2.43 (m, 2 H–C(2)(G₁)); 2.98 (d, J=5.2, H–C(4)(C)); 3.35–4.75 (m, 6 CH₂OAllyl, 2 OCH₂-C(2)(G₁), CH₂(C), 6 H–C(3)(G₂), 18 CH₂=CHCH₂, 2 H–C(3)(G₁), H–C(2)(C), 8 OCH₂Ph, 6 H–C(1)(G₂), 2 H–C(1)(G₁)); 5.05–5.35 (m, 18 CH₂=CHCH₂); 5.63–5.97 (m, 18 CH₂=CHCH₂); 7.17–7.37 (m, 32 arom. H). ¹³C-NMR (125 MHz): 14.52; 16.96; 26.49; 35.76; 44.98; 46.12; 46.33; 65.90; 66.46; 68.04; 69.99; 70.77; 70.98; 71.75; 72.71; 73.12; 73.89; 74.24; 74.72; 75.21; 75.45; 78.82; 78.85; 79.52; 80.08; 116.40; 116.44; 116.80; 116.84; 117.49; 117.52; 126.05; 126.75; 127.21; 127.30; 127.33; 127.38; 127.41; 127.47; 127.69; 127.78; 127.80; 127.87; 129.02; 134.14; 134.16; 134.18; 134.45; 134.48; 134.54; 134.73; 134.77; 137.35; 137.69; 138.00; 138.05; 138.50; 139.10; 139.19; 139.25; 139.57; 139.66. ¹⁹F-NMR (282 MHz): -74.30 to -74.51 (m, 8 CF₃) MALDI-TOF: 3017 ([M + Na]⁺).

 $(MeO)_0$ -{ $\{(S)G_1(F_9)(Biph.)\}^3\}_3$ -{ $C\}$ } (38). As described for 33, with 6 (50 mg, 0.28 mmol), NaH (62 mg, 2.55 mmol), and 28 (524 mg, 1.14 mmol), reaction time 4 h. FC (pentane/Et₂O 5:1 → 2:1): 38 (138 mg, 50%). ¹H-NMR (200 MHz): 1.00 (s, t-Bu); 1.37 (d, J = 6.2, Me); 2.19-2.24 (m, 3 H-C(2)(G_1)); 2.36-2.38 (m, H-C(3)(C)); 3.13 (2s, 3 MeO(P)); 3.27 (3s, 3 MeO(P)); 3.28-3.38 (m, 3 CH₂OMe(P), H-C(4)(C)); 3.60-3.70 (m, 1 H, CH₂(C), 3 MeO(P)); 3.81-3.90 (m, 1 H, CH₂(C), H-C(2)(C)); 4.10-4.19 (m, 3 H-C(3)(G_1)); 4.32-4.74 (m, 3 H-C(1)(G_1), 3 OCH₂Ph); 7.26-7.68 (m, 24 arom. H). ¹⁹F-NMR (282 MHz): -73.69 to -73.71 (br. t, 3 CF₃).

 $(MeO)_0 - \{[(S)G_1(F_0)(Biph.)]^3\}_3 - [(C_{enant.})]$ (39). As described for 33, with *ent-*6 (50 mg, 0.28 mmol), NaH (62 mg, 2.55 mmol), and 28 (524 mg, 1.14 mmol), reaction time 4 h. FC (pentane/Et₂O 4:1 \rightarrow 1:1): 39 (150 mg, 53%). [α]₀^{r.t.} = -38.0 (c = 1.4). IR: 3005w, 2934m, 1497w, 1463w, 1383w, 1278m, 1169s, 1139s, 1102s. ¹H-NMR (200 MHz): 1.02 (s, t-Bu); 1.28 (d, J = 6.2, Me); 2.15 - 2.25 (m, 3 H – C(2)(G_1)); 2.27 - 2.42 (m, H – C(3)(C)); 3.15

 $(MeO)_{27} - \{[(S)G_2(F_{36})(Biph.)]^3\}_3 - \{C\} \ (\textbf{40}). \text{ As described for } \textbf{33}, \text{ with } \textbf{6} \ (33 \text{ mg}, 0.18 \text{ mmol}), \text{ NaH } (41 \text{ mg}, 1.71 \text{ mmol}), \text{ and } \textbf{32} \ (1.0 \text{ g}, 0.67 \text{ mmol}), \text{ reaction time } \textbf{4} \text{ h. } \text{FC} \ (\text{pentane/Et}_2O \ 1:1): \ \textbf{40} \ (200 \text{ mg}, 25\%). \\ [2]_D^{\text{Li}} = -45.8 \ (c = 1.0). \text{ IR}: 2985w, 2933m, 1492w, 1461w, 1390m, 1277m, 1174s, 1138s, 1097s, 1010w. }^{1} \text{H-NMR} \ (500 \text{ MHz}): 1.01 \ (s, t\text{-Bu}); \ 1.38 \ (d, J = 6.3, \text{ Me}); \ 2.18 - 2.19 \ (m, 9 \text{ H} - \text{C(2)}(\text{G}_2)); \ 2.42 - 2.49 \ (m, \text{H} - \text{C(3)}(\text{C}), 3 \text{ H} - \text{C(2)}(\text{G}_1)); \ 3.09 - 3.11 \ (3s, 9 \text{ MeO(P)}); \ 3.24 - 3.26 \ (4s, 9 \text{ MeO(P)}); \ 3.29 - 3.36 \ (m, 9 \text{ CH}_2\text{OMe(P)}, \text{H} - \text{C(4)}(\text{C)}); 3.61 - 3.69 \ (3s, 9 \text{ MeO(P)}, 3 \text{ OCH}_2 - \text{C(2)}(\text{G}_1), 1 \text{ H, CH}_2(\text{C)}); 3.82 - 3.87 \ (m, \text{H} - \text{C(2)}(\text{C)}); 3.90 - 3.93 \ (m, 1 \text{ H, CH}_2(\text{C})); 4.09 - 4.16 \ (m, 9 \text{ H} - \text{C(3)}(\text{G}_2)); 4.23 - 4.79 \ (m, 3 \text{ H} - \text{C(3)}(\text{G}_1), 9 \text{ H} - \text{C(1)}(\text{G}_2), 3 \text{ H} - \text{C(1)}(\text{G}_1), 12 \text{ OCH}_2\text{Ph}); 7.20 - 7.47 \ (m, 96 \text{ arom. H}). \\ ^{13}\text{C-NMR} \ (125 \text{ MHz}): 16.60; 26.44; 26.57; 29.71; 37.55; 46.05; 46.39; 56.88; 58.65; 61.29; 66.47; 68.21; 68.90; 70.64; 71.04; 72.74; 72.85; 73.66; 74.75; 74.99; 75.08; 75.31; 75.45; 76.76; 80.00; 80.81; 86.08; 124.64; 126.74; 126.91; 126.94; 127.00; 127.03; 127.04; 127.06; 127.15; 127.22; 127.42; 127.81; 127.86; 127.97; 128.00; 128.21; 128.32; 128.64; 129.23; 136.76; 137.14; 137.54; 138.23; 138.34; 138.71; 138.75; 138.79; 138.83; 138.92; 139.45; 139.75; 139.79; 140.26; 140.32; 140.45; 140.62; 140.66; 140.69. \\ ^{19}\text{F-NMR} \ (282 \text{ MHz}): -73.70 \text{ to } -73.79 \ (m, 3 \text{ CF}_3(\text{G}_2)); -74.09 \text{ to } -74.16 \ (m, \text{ CF}_3(\text{G}_1)). \text{ MALDI-TOF: } 4637 \ ([M+Na]^+). \\ \end{array}$

X-Ray Crystal-Structure Analysis of Compound 7 ($C_{35}H_{41}O_6F_3Si$). Determination of the cell parameters and collection of the reflection intensities were performed on an Enraf-Nonius-CAD4 four-circle diffractometer (graphite monochromatized Mo K_a radiation, $\lambda=0.7107$ Å). Colorless prism, $0.4\times0.6\times0.8$ mm, monoclinic, space group P_{21} , a=10.960(3) Å, b=8.980(5) Å, c=18.396(4) Å, $\beta=103.48(2)$, V=1067(7) ų, Z=2, $\rho_{calc}=1.212$ gcm⁻³, $\mu=0.124$ mm⁻¹, F(000)=680. Number of reflections measured 3294 ($\omega/20$ scan, $2<20<40^\circ$, T=295 K); 3294 unique reflections, which were used for the determination (direct methods, SHELXS-86). SHELXS-93 was used for structure refinement (full-matrix least squares). The non-H-atoms were refined anisotropically, the H-atoms were added to the molecule with constant isotropic temp, factors on idealized positions and refined according to the riding model (afix 3). The refinement converged at R=0.041 ($wR^2=0.099$), min. and max. rest electron density -0.19, 0.15 eÅ⁻³, number of variables 406.

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